# Synthesis of Oligomeric 4-(Glycosyloxy)benzoate Macrocyclic Glycosides 

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(S) Supporting Information


#### Abstract

Clemoarmanoside A and Clemahexapetoside A, two novel cyclic dimers of 4-(glycosyloxy)benzoates containing the unusual D-allopyranose as one of the sugar units, were synthesized for the first time. The convenient synthetic approach was adapted to the assembly of the symmetrical trimeric, tetrameric, and pentameric congeners. The synthesis clarified the discrepancy in the NMR data reported for the natural products. X-ray diffraction analysis of Clemahexapetoside A revealed that it adopted an armchair conformation with two carbohydrate rings as the arms and two aromatic rings as the back and seat, respectively.


Clemoarmanoside $A(R=H)$
Clemahexapetoside

Clemahexapetoside $A(R=O M e)$

## INTRODUCTION

Since 1992, eight cyclic dimers of 4-(glycosyloxy)benzoates have been identified from a variety of the folk medicinal plants (Table 1). ${ }^{1-8}$ Compounds 1, 2, 5, and 6 are homodimers, while compounds $3,4,7$, and 8 consist of two slightly different monomers.

The cyclic structure of these natural metabolites is reminiscent of the artificial glycophanes, ${ }^{9}$ which possess a chiral cavity with both hydrophilic and hydrophobic facets and thus could be utilized as a new type of host molecules. In addition, these scarce components of medicinal plants might possess various bioactivities. Thus, synthesis of these molecules and their congeners becomes an interesting subject. The homodimer $\mathbf{1}$ has been synthesized via cyclodimerization of $4-O-(2,3,4$-tri-O-benzyl- $\beta$-D-glucopyranosyl) syringic acid. ${ }^{10}$ Recently, compounds $1-3$ were synthesized in a one-pot fashion via homo- and heterocyclodimerization of two types of the monomers using the fluorous mixture synthesis strategy. ${ }^{11}$

Different from other members, compounds 7 and 8, named Clemoarmanoside A and Clemahexapetoside A, respectively, contain the unusual D-allopyranose as one of the sugar units. Their synthesis would require coupling of two disparate monomers prior to cyclization. The great discrepancy in the reported ${ }^{1}$ H NMR data for Clemoarmanoside A (7) from two research groups was an additional impetus for us to embark on their synthesis. ${ }^{5,6}$ Herein, we report the synthesis of these two novel cyclic glycosides, the X-ray diffraction analysis of Clemahexapetoside A (8), and the extension of the present synthetic approach to the preparation of trimeric, tetrameric, and pentameric congeners 9-11.


## RESULTS AND DISCUSSION

Synthesis of Clemoarmanoside A (7) and Clemahexapetoside A (8). To synthesize the cyclic heterodimers Clemoarmanoside A (7) and Clemahexapetoside A (8), three 4-(glycosyloxy) benzoate building blocks (i.e., 17, 26, and 27) were designed, in that benzyl ethers were used as permanent protecting groups while trityl ether and allyl ester, which are selectively removable after ester condensation, were used as temporary protecting groups. The preparation of these 4-(glycosyloxy)benzoate derivatives were straightforward, as shown in Schemes 1 and 2.

Thus, glycosylation of methyl vanillate with peracetyl glucopyranosyl trifluoroacetimidate 12 under the action of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ (0.1 equiv) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0{ }^{\circ} \mathrm{C}$ afforded the phenolic $\beta$-dglucoside $13(\mathrm{H}-1,5.07 \mathrm{ppm}, J=7.5 \mathrm{~Hz})$ in a high $90 \%$ yield without detection of the corresponding $\alpha$-anomer (Scheme 1 ). ${ }^{12}$ Worth noting is that replacement of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ with TMSOTf as the promoter in the present glycosylation reaction led to a complex mixture. Removal of the acetyl groups $\left(\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{MeOH}\right.$,

[^0]
## Table 1. Naturally Occurring Cyclic Dimers of 4-(Glycosyloxy)benzoates





| compounds | plants and references |
| :---: | :---: |
| 1 (Clemochinenoside A): $\mathrm{R}=\mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{OMe}$ | Clematis chinensis, ${ }^{1}$ C. hexapetala, ${ }^{6}$ C. hexapetala (Ranunculaceae), ${ }^{7}$ and Capparis tenera (Capparidaceae) ${ }^{8}$ |
| 2 (Berchemolide): $\mathrm{R}=\mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=\mathrm{OMe}$ | Berchemia racemosa (Rhamnaceae) ${ }^{2}$ and Clematis armandii (Ranunculaceae) ${ }^{5}$ |
| 3 (Clemochinenoside B): $\mathrm{R}=\mathrm{H}, \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{OMe}$ | Clematis chinensis, ${ }^{3}$ C. armandii, ${ }^{5}$ C. hexapetala (Ranunculaceae), ${ }^{7}$ and Capparis tenera (Capparidaceae) ${ }^{8}$ |
| 4 (Clemoarmanoside B): $\mathrm{R}=\mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{H}$ | Clematis armandii (Ranunculaceae) ${ }^{5}$ |
| 5 (Pakistolide A ): $\mathrm{R}=\beta$-D-glucopyranosyl | Berchemia pakistanica (Rhamnaceae) ${ }^{4}$ |
| 6 (Pakistolide B): $\mathrm{R}=(4 Z)$-1,6-dioxohept-4-enyl | Berchemia pakistanica (Rhamnaceae) ${ }^{4}$ |
| 7 (Clemoarmanoside A ): $\mathrm{R}=\mathrm{R}^{1}=\mathrm{H}$ | Clematis armandii ${ }^{5}$ and C. hexapetala (Ranunculaceae) ${ }^{6}$ |
| 8 (Clemahexapetoside A ): $\mathrm{R}=\mathrm{H}, \mathrm{R}^{1}=\mathrm{OMe}$ | Clematis hexapetala (Ranunculaceae) ${ }^{6}$ |

Scheme 1. Preparation of 4-(Glucopyranosyloxy)benzoate Derivative 17





THF, rt, $95 \%$ ) led to tetraol $\mathbf{1 4},{ }^{11}$ which was subjected to selective protection of the primary $6-\mathrm{OH}$ with trityl group ( TrCl , pyridine, DMAP, $55^{\circ} \mathrm{C}$ ) followed by protection of the remaining $2,3,4-\mathrm{OHs}$ with benzyl group to provide 15 ( $76 \%$ for two steps). The trityl group on 15 was removed with $90 \%$
$\mathrm{CF}_{3} \mathrm{COOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0^{\circ} \mathrm{C}$ to give $16(91 \%) .{ }^{13}$ Conversion of the methyl ester in 16 into allyl ester via hydrolysis ( $\mathrm{KOH}, \mathrm{THF}$, $\mathrm{H}_{2} \mathrm{O}$, reflux) and subsequent allyl ester formation (AllBr, $\mathrm{K}_{2}$ $\mathrm{CO}_{3}$, DMF, $\left.50^{\circ} \mathrm{C}, 79 \%\right)^{14}$ furnished the desired 4-(glycosyloxy) benzoate building block 17 .

Scheme 2. Preparation of 4-(Allopyranosyloxy)benzoate Derivatives 26 and 27


1,2,3,4,6-Penta-O-acetyl-D-allopyranose $\mathbf{1 8}$ was prepared from D-glucose ( 5 steps, $32 \%$ overall yield) following a literature method (Scheme 2). ${ }^{15}$ Selective removal of the anomeric acetyl group $\left(\mathrm{BnNH}_{2}, \mathrm{THF}, \mathrm{rt}\right)$ followed by trifluoroacetimidate formation $\left(\mathrm{CF}_{3}(\mathrm{Cl}) \mathrm{C}=\mathrm{NPh}, \mathrm{K}_{2} \mathrm{CO}_{3} \text {, acetone, } \mathrm{rt} \text {; } 82 \%\right)^{12}$ provided allopyranosyl donor 19. Glycosylation of methyl vanillate with 19 under the effect of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ ( 0.1 equiv) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0^{\circ} \mathrm{C}$ provided the desired $\beta$-D-allopyranoside $20(\mathrm{H}-1,5.38 \mathrm{ppm}, J=8.4 \mathrm{~Hz})$ nearly quantitatively. Similar conditions applied to the coupling of 19 with the stereohindered methyl syringate led to $\beta$-D-allopyranoside $21(\mathrm{H}-1,5.32 \mathrm{ppm}, J=8.1 \mathrm{~Hz})$ also in good yield (87\%). Employing similar protecting group manipulations as that in the synthesis of glucoside 17 (Scheme 1), allosides 20 and 21 were readily converted into the desired 4-(allopyranosyloxy)benzoate building blocks 26 and 27 via four convenient steps, that is, deacetylation, selective tritylation, benzylation, and hydrolysis of methyl ester.

With building blocks 17, 26, and 27 in hand, the stage was set for final assembly of the target macrocyclic glycosides 7 and $\mathbf{8}$ (Scheme 3). Treatment with DCC and DMAP allowed alcohol $\mathbf{1 7}$ to be coupled with acid 26 or 27 to form ester 28 or 29 in $\sim 75 \%$ yields. Removal of the 6-O-trityl group from 28 and 29 ( $90 \% \mathrm{CF}_{3} \mathrm{COOH}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}$ ) followed by cleavage of the allyl ester $\left(\mathrm{Pd}^{\left.\left(\mathrm{PPh}_{3}\right)_{4}, \mathrm{Et}_{2} \mathrm{NH}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}\right)^{16} \text { afforded the key }}\right.$ intermediates 32 and 33 in satisfactory yields. Intramolecular esterification of 32 and 33 proceeded sluggishly under the action of DCC and DMAP, leading to the desired cyclic esters 34 and 35 in $<10 \%$ yields. Nevertheless, under the Yamaguchi conditions (2,4,6-trichlorobenzoyl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DMAP}$ ) ${ }^{17}$ in a diluted toluene solution $\left(2.5 \times 10^{-3} \mathrm{~mol} / \mathrm{L}\right), 32$ and 33 were converted into cyclic esters 34 and 35 in satisfactory yields ( 78 and $88 \%$, respectively); polar byproducts were discerned on TLC, which were presumably derived from the intermolecular condensation. Finally, global removal of the benzyl groups on 34 and 35 was effected via hydrogenolysis over $\mathrm{Pd}(\mathrm{OH})_{2}$, furnishing the target macrocyclic glycosides 7 and 8 in 81 and $100 \%$ yield, respectively.

The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data of the synthetic compound 7 were identical to those of the natural product Clemoarmanoside A reported by Xu et al. ${ }^{5}$ Identical ${ }^{13} \mathrm{C}$ NMR data were

Scheme 3. Assembly of the Macrocyclic Glycosides 7 and 8

reported by Tu et al. for the same natural product; however, the ${ }^{1} \mathrm{H}$ NMR data Tu et al. provided were in great discrepancy. ${ }^{6}$ The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data of the natural compound Clemahexapetoside A reported by Tu et al. were also in apparent discrepancy with those of the synthetic compound 8. ${ }^{6}$ Fortunately, a single crystal of compound 8 was obtained, and the X-ray diffraction analysis confirmed the correctness of the synthetic structure. ${ }^{18}$

X-ray Diffraction Analysis of Clemahexapetoside A (8). Clemahexapetoside A (8) crystallized in the monoclinic crystal system with space group P21. Each unit cell of the crystal structure contained two identical macrocycles and five water molecules (Figures 1 and 2), with one water molecule disordered. The whole structure exhibited as an armchair with two carbohydrate rings as the arms and two aromatic rings ( $\mathrm{A}, \mathrm{C} 8-\mathrm{C} 13 ; \mathrm{B}, \mathrm{C} 22-\mathrm{C} 27$ ) as the back and seat, respectively. The stability of the crystal structure resulted from dense networks of hydrogen bonds and hydrophobic interactions. The intramolecular hydrophobic interactions were observed between methyl group C29 and aromatic ring A. Intermolecular hydrophobic interactions mainly occurred between methyl C14 and aromatic ring B of the adjacent molecule. Hydroxyl groups of the two sugar units and H atoms of the five water molecules formed strong hydrogen bondings, which played a major role in the dense stacking of the crystal structure (see Supporting Information).

Synthesis of the Cyclic Trimer, Tetramer, and Pentamer $9-11$. The above synthetic approach to natural cyclic dimers of 4-(glycosyloxy)benzoates was successfully adapted to the synthesis of larger cyclic congeners (i.e., 9-11). Cyclic trimer 9, which possesses a $C_{3}$ symmetry, was assembled starting from $\beta$-Dglucoside 36 (5 steps and $54 \%$ overall yield from D-glucose) ${ }^{19}$
(Scheme 4). Thus, selective tritylation of the primary 6-OH on 36 followed by benzylation of the remaining OHs provided 37 ( $70 \%$ for 2 steps). Hydrolysis of the methyl ester on 37 gave acid 38 (91\%). Removal of the trityl ether on 38 followed by chemoselective formation of the allyl ester furnished alcohol 40 ( $65 \%$ for 2 steps). Condensation of acid 38 and alcohol 40 under the action of EDCI and DMAP led to the dimer 41 in a satisfactory $87 \%$ yield. Under similar conditions, the corresponding alcohol resulting from 41 via removal of the trityl group was condensed with acid 38 to provide the linear trimer 42 in $61 \%$ yield (2 steps). The $90 \% \mathrm{CF}_{3} \mathrm{COOH}$ mediated detritylation of 42 led to 43 ( $85 \%$ ). The allyl ester on 43 was cleaved smoothly with $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}\left(\mathrm{Et}_{2} \mathrm{NH}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}, 75 \%\right)$, and the resulting alcohol acid was subjected to the Yamaguchi conditions to afford the cyclic trimer 44 in a good $64 \%$ yield. Global cleavage of the benzyl groups via hydrogenolysis furnished the desired trimer 9 quantitatively.

Following a similar reaction sequence as in the trimer synthesis, tetramer $\mathbf{1 0}$ and pentamer $\mathbf{1 1}$ were successfully synthesized (Scheme 5). The intramolecular macrolactonization of the linear tetramer and pentamer under Yamaguchi conditions met with no


Figure 1. View of molecule 8 showing the atom labeling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level for non- H atoms.
problem, leading to the corresponding macrocyclic glycosides 47 and 50 in 60 and $50 \%$ yield, respectively. Interestingly, the final tetrameric cyclic glycoside $\mathbf{1 0}$ was hardly soluble in methanol, while the trimeric and pentameric congeners 9 and 11 were easily soluble. The ${ }^{1} \mathrm{H}$ NMR of the cyclic oligomers $9-11$ showed only one set of the signals corresponding to the monomer 4-(glycosyloxy) benzoate, indicating that a symmetrical conformation was adopted by these macrocyclic glycosides in solution.

In summary, two natural cyclic dimers of 4-(glycosyloxy) benzoates containing the unusual D-allopyranose as one of the sugar units, Clemoarmanoside A (7) and Clemahexapetoside A (8), were synthesized for the first time. The synthesis clarified the discrepancy in the NMR data reported for these natural products. X-ray diffraction analysis of Clemahexapetoside A (8) revealed that it adopted an armchair conformation with two carbohydrate rings as the arms and two aromatic rings as the back and seat, respectively. The macrocyclic trimer, tetramer, and pentamer ( $9-11$ ) were also assembled conveniently with the present synthetic approach. The inclusion properties of these macrocyclic glycosides as a new type of glycophanes will be studied, and the results will be reported in due course.

## ■ EXPERIMENTAL SECTION

Methyl 3-Methoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}$-tri-O-benzyl-6'-O-trityl- $\beta$-dglucopyranosyloxy)benzoate (15). To a solution of glucoside $14(1.7 \mathrm{~g}, 5.1 \mathrm{mmol})$ in pyridine $(40 \mathrm{~mL})$ were added $\operatorname{TrCl}(5.6 \mathrm{~g}, 20.0$ mmol ) and DMAP ( $62 \mathrm{mg}, 0.5 \mathrm{mmol}$ ). After stirring at $55^{\circ} \mathrm{C}$ for 5 h , the mixture was concentrated under reduced pressure. The residue was purified by silica gel column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}, 40: 1\right)$ to provide methyl 3 -methoxy-4-(6' ${ }^{\prime}$ - - -trityl $-\beta$-D-glucopyranosyloxy)benzoate ( $2.46 \mathrm{~g}, 83 \%$ ) as a yellow solid.

The above solid was dissolved in dry DMF ( 40 mL ), the solution was cooled to $0^{\circ} \mathrm{C}$, to which $\mathrm{NaH}(60 \%, 670 \mathrm{mg}, 28.0 \mathrm{mmol})$ was added portionwise. After stirring at $0^{\circ} \mathrm{C}$ for $15 \mathrm{~min}, \mathrm{BnBr}(3.5 \mathrm{~mL}, 25.2 \mathrm{mmol})$ was added. The resulting mixture was stirred at room temperature for a further 3 h before $\mathrm{MeOH}(15 \mathrm{~mL})$ was added to quench the reaction. Solvent removal gave a residue, which was purified by column chromatography on silica gel (petroleum ether/EtOAc, 6:1) to give 15 ( 3.59 g , $91 \%$ ) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-21.7\left(c 0.4, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.68-7.64(\mathrm{~m}, 2 \mathrm{H}), 7.48-7.19(\mathrm{~m}, 30 \mathrm{H}), 6.85(\mathrm{~m}, 2 \mathrm{H})$, $5.30(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.08(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.98-4.80(\mathrm{~m}, 3 \mathrm{H})$,


Figure 2. Molecular packing of macroglycoside 8 viewed along the $b$ axis. Dashed lines indicate the hydrogen bonding interactions. Thermal ellipsoids are drawn on the $50 \%$ probability level.

Scheme 4. Synthesis of the Cyclic Trimer of 4-(Glycosyloxy)benzoate 9

$4.72(\mathrm{~d}, J=10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.35(\mathrm{~d}, J=10.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.91-3.81(\mathrm{~m}, 7 \mathrm{H})$, $3.78-3.58(\mathrm{~m}, 4 \mathrm{H}), 3.32(\mathrm{dd}, J=4.8,9.9 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 166.8,150.8,149.4,143.8,138.5,137.6,128.8,128.5,128.4$ (2 C), 128.3, 128.2, 128.0, 127.9, 127.8, 127.0, 124.7, 123.3, 116.0, 112.7, 102.2, 86.6, 84.4, 81.9, 77.8, 76.0, 75.2, 75.0, 74.9, 62.4, 55.9, 52.1; HRMS (MALDI) calcd for $\mathrm{C}_{55} \mathrm{H}_{52} \mathrm{O}_{9} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$879.3504, found 879.3500.

Methyl 3-Methoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}$-tri-O-benzyl- $\beta$-d-glucopyranosyloxy)benzoate (16). To a solution of trityl ether 15 ( 171 mg , $0.2 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added dropwise a $90 \%$ $\mathrm{CF}_{3} \mathrm{COOH}$ solution $(0.02 \mathrm{~mL})$. After stirring at $0{ }^{\circ} \mathrm{C}$ for 30 min , the solution was concentrated under vacuum. The residue was purified by column chromatography on silica gel (petroleum ether/EtOAc, 4:1) to give 16 ( $123 \mathrm{mg}, 91 \%$ yield) as a white solid: $[\alpha]_{\mathrm{D}}^{25}=-23.9$ (c 0.2, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.66-7.59(\mathrm{~m}, 2 \mathrm{H})$, $7.34-7.26(\mathrm{~m}, 15 \mathrm{H}), 7.07(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{~d}, J=10.5 \mathrm{~Hz}$, $1 \mathrm{H}), 5.12-5.10(\mathrm{~m}, 1 \mathrm{H}), 5.01(\mathrm{~d}, J=11.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.91-4.81(\mathrm{~m}, 3$ H), $4.68(\mathrm{~d}, J=11.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-3.89(\mathrm{~m}, 1 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 3.89(\mathrm{~s}$, $3 \mathrm{H}), 3.79-3.63(\mathrm{~m}, 4 \mathrm{H}), 3.55-3.52(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 166.6,150.3,149.3,138.4,138.2,137.8,128.5,128.4,128.3$ (2 C), 128.0 ( 2 C ), 127.8 ( 2 C ), 127.7, 124.9, 123.1, 115.1, 112.8, 101.7, 84.2, 81.6, 77.1, 75.7, 75.5, 75.1, 74.8, 61.9, 55.9, 52.1; HRMS (MALDI) calcd for $\mathrm{C}_{36} \mathrm{H}_{38} \mathrm{O}_{9} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$637.2408, found 637.2413.

Allyl 3-Methoxy-4-(2', $3^{\prime}, 4^{\prime}$-tri-O-benzyl- $\boldsymbol{\beta}$-d-glucopyranosyloxy)benzoate (17). To a solution of methyl ester 16 ( 115 mg , 0.19 mmol ) in THF ( 10 mL ) was added dropwise a solution of KOH $(319 \mathrm{mg}, 5.70 \mathrm{mmol})$ in water $(1 \mathrm{~mL})$. The mixture was heated to $60^{\circ} \mathrm{C}$, and the stirring was continued for $10 \mathrm{~h} . \mathrm{HOAc}$ was added dropwise to adjust the pH value of the reaction mixture to 7.0. The resulting mixture was extracted with ethyl acetate $(100 \mathrm{~mL} \times 3)$. The organic layers were combined, washed with saturated NaCl , dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and then concentrated under vacuum to give a crude product.

The crude product was dissolved in DMF $(15 \mathrm{~mL})$, to which AllBr $(0.1 \mathrm{~mL}, 1 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(2 \mathrm{mg}, 0.01 \mathrm{mmol})$ were added. The resulting mixture was stirred for 5 h at $50^{\circ} \mathrm{C}$ and was then concentrated under vacuum. The residue was purified by column chromatography on silica gel (petroleum ether/EtOAc, 4:1) to give compound 17 ( 95 mg , $79 \%$ for two steps) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-12.5\left(c 0.7, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.69(\mathrm{dd}, J=4.2,8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~d}, J=$ $1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.34-7.28(\mathrm{~m}, 15 \mathrm{H}), 7.07(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.11-5.98$ $(\mathrm{m}, 1 \mathrm{H}), 5.44(\mathrm{~d}, J=14.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.31(\mathrm{~d}, J=10.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{~d}$, $J=11.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.12-5.10(\mathrm{~m}, 1 \mathrm{H}), 5.01(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H})$, $4.91-4.81(\mathrm{~m}, 5 \mathrm{H}), 4.68(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-3.92(\mathrm{~m}, 1 \mathrm{H}), 3.89$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 3.79-3.63 (m, 4 H$), 3.56-3.51(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 165.8,150.4,149.3,138.4,138.2,137.8,132.3,128.5$,

Scheme 5. Synthesis of the Cyclic Tetramer and Pentamer of 4-(Glycosyloxy)benzoates 10 and 11



$\mathrm{Pd}(\mathrm{OH})_{2}, \mathrm{H}_{2}(1 \mathrm{~atm}), \square \mathbf{4 7}$
$\mathrm{THF} / \mathrm{H}_{2} \mathrm{O}, \mathrm{rt}, 100 \%$ $\mathbf{1 0} 0$

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$\mathrm{Pd}(\mathrm{OH})_{2}, \mathrm{H}_{2}(1 \mathrm{~atm}), \mathrm{THF} / \mathrm{H}_{2} \mathrm{O}, \mathrm{rt}, 100 \% \square \mathbf{5 0}$
128.4 (2 C), 128.3, 128.0 (2 C), 127.8 (2 C), 127.7, 124.8, 123.2, 118.2, 115.1, 112.9, 101.6, 84.2, 81.6, 77.1, 75.7, 75.5, 75.1, 74.8, 65.5, 61.9, 55.9; HRMS (MALDI) calcd for $\mathrm{C}_{38} \mathrm{H}_{40} \mathrm{O}_{9} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 663.2565$, found 663.2563 .

2,3,4,6-Tetra-O-acetyl-d-allopyranosyl $N$-phenyltrifluoroacetimidate (19). To a solution of 2,3,4,6-tetra-O-acetyl-D-allopyranose ( $100 \mathrm{mg}, 0.29 \mathrm{mmol}$ ) in acetone $(5 \mathrm{~mL})$ were added trifluoroacetimidoyl chloride ( $89 \mathrm{mg}, 0.43 \mathrm{mmol}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(60 \mathrm{mg}, 0.43 \mathrm{mmol})$. After being stirred at room temperature for 3 h , the solid was removed by filtration and the filtrate was evaporated under reduced pressure. The residue was purified by silica gel column chromatography (petroleum ether/EtOAc, 2:1) to give $19(119 \mathrm{mg}, 82 \%)$ as a syrup. A small portion of the $\beta$ isomer was isolated for characterization: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.37(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H})$, $7.18(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.90(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.06(\mathrm{~s}, 1 \mathrm{H}), 5.73(\mathrm{~s}, 1 \mathrm{H})$, $5.14-5.03(\mathrm{~m}, 2 \mathrm{H}), 4.22-4.11(\mathrm{~m}, 3 \mathrm{H}), 2.17(\mathrm{~s}, 3 \mathrm{H}), 2.09(\mathrm{~s}, 3 \mathrm{H}), 2.08$ $(\mathrm{s}, 3 \mathrm{H}), 2.01(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.4,169.9,169.2$, 168.8, 142.9, 128.8, 124.6, 119.1, 94.5, 72.6, 72.4, 70.2, 67.7, 61.4, 20.4, 20.3, 20.2; HRMS (MALDI) calcd for $\mathrm{C}_{22} \mathrm{H}_{24} \mathrm{~F}_{3} \mathrm{NO}_{10} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$542.1245, found 542.1238 .

Methyl 3-Methoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetra- $O$-acetyl- $\beta$-d-allopyranosyloxy)benzoate (20). To a mixture of allosyl trifluoroacetimidate $19(6.23 \mathrm{~g}, 12.63 \mathrm{mmol})$, methyl vanillate $(1.93 \mathrm{~g}, 10.53 \mathrm{mmol})$, and $4 \AA \mathrm{MS}$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}(0.14 \mathrm{~mL}, 1.30 \mathrm{mmol})$ dropwise at $0{ }^{\circ} \mathrm{C}$. Stirring was continued for further 2 h at the same temperature. The MS was removed via filtration through a pad of Celite. The filtrate was concentrated under vacuum to give a residue, which was purified by column chromatography on silica gel (petroleum ether/EtOAc, 2.5:1) to afford $20(5.34 \mathrm{~g}, 99 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=-13.3\left(c 1.7, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.65(\mathrm{dd}, J=1.5,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~d}, J=1.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.17(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.77(\mathrm{t}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.38(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.27(\mathrm{dd}, J=3.0,8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.08-5.03(\mathrm{~m}, 1 \mathrm{H}), 4.26-4.22(\mathrm{~m}$, $3 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 2.17(\mathrm{~s}, 3 \mathrm{H}), 2.09(\mathrm{~s}, 3 \mathrm{H}), 2.06(\mathrm{~s}, 3 \mathrm{H})$, $2.03(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 172.3,171.4,170.8,170.7$, 168.2, 152.0, 151.6, 127.3, 124.7, 118.6, 115.2, 99.7, 72.4, 70.3, 70.2, 67.9, 64.0, 57.9, 53.8, 22.4 (2C), 22.2 (2 C); HRMS (MALDI) calcd for $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{O}_{13} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 535.1422$, found 535.1434.

Methyl 3,5-Dimethoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetra- $O$-acetyl- $\beta$-d-allopyranosyloxy)benzoate (21). Similar procedure as that used for the synthesis of alloside 20 (from 19) was applied to provide 21 ( 800 mg , $87 \%$ from 19) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-7.1\left(c 1.9, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.23(\mathrm{~s}, 2 \mathrm{H}), 5.65(\mathrm{t}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.32(\mathrm{~d}, J=8.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.17(\mathrm{dd}, J=3.0,8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.04(\mathrm{dd}, J=5.4,9.6 \mathrm{~Hz}, 1 \mathrm{H})$, 4.17-4.15 (m, 2 H), 4.06-4.01 (m, 1 H), 3.84 ( $\mathrm{s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 6 \mathrm{H}), 2.07$ ( s, 3 H$), 1.99(\mathrm{~s}, 3 \mathrm{H}), 1.94(\mathrm{~s}, 3 \mathrm{H}), 1.91(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 170.6,169.7,169.1,169.0,166.4,152.6,138.6,126.2,107.1,99.5$, 70.9, 69.6, 68.4, 66.2, 62.4, 56.4, 52.3, 21.0, 20.7 (2C), 20.5; HRMS (MALDI) calcd for $\mathrm{C}_{24} \mathrm{H}_{30} \mathrm{O}_{14} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$565.1528, found 565.1525.

Methyl 3-Methoxy-4-( $\beta$-d-allopyranosyloxy)benzoate (22). To a solution of $20(200 \mathrm{mg}, 0.39 \mathrm{mmol})$ in THF/MeOH $(10 \mathrm{~mL}, \mathrm{v} / \mathrm{v}=$ $1: 1)$ was added $\mathrm{K}_{2} \mathrm{CO}_{3}(6 \mathrm{mg}, 0.04 \mathrm{mmol})$. The resulting suspension was stirred at room temperature for 2 h and was then filtered through a pad of Celite. The filtrate was concentrated to give a residue, which was subjected to column chromatography on silica gel $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}, 10: 1\right)$ to give $22(122 \mathrm{mg}, 91 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-33.7(c 2.5, \mathrm{MeOH}) ;{ }^{1} \mathrm{H}$ NMR ( $\left.300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 7.67(\mathrm{dd}, J=2.1,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~d}, J=$ $1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.37(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.16(\mathrm{t}, J=$ $2.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 3.87(\mathrm{~s}, 3 \mathrm{H}), 3.87-3.85(\mathrm{~m}, 2 \mathrm{H}), 3.70-3.58$ $(\mathrm{m}, 3 \mathrm{H}), 3.31(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta$ 166.8, 151.0, 148.9, 123.7, 123.1, 114.8, 112.6, 98.5, 74.4, 71.5, 70.4, 67.1, 61.3, 55.2, 51.1; HRMS (ESI) calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}_{9} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 367.1000, found 367.1001.

Methyl 3,5-Dimethoxy-4-( $\boldsymbol{\beta}$-d-allopyranosyloxy)benzoate (23). Similar procedure as that for $20 \rightarrow 22$ was used to convert 21 into 23 $(60 \mathrm{mg}, 99 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=-15.8(c 0.5, \mathrm{MeOH}) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 7.35(\mathrm{~s}, 2 \mathrm{H}), 5.38(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{~s}, 1 \mathrm{H})$, $3.90(\mathrm{~s}, 9 \mathrm{H}), 3.78(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.67-3.62(\mathrm{~m}, 4 \mathrm{H}), 3.31(\mathrm{~s}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta$ 168.1, 154.2, 140.7, 127.0, 108.4, 103.1, 76.4, 73.1, 72.4, 68.7, 63.0, 57.1, 52.8; HRMS (MALDI) calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{10} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$397.1105, found 397.1106.

Methyl 3-Methoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}$-tri-O-benzyl-6'-O-trityl- $\beta$-dallopyranosyloxy)benzoate (24). Similar procedure as that used for $14 \rightarrow 15$ was adopted to convert tetraol 22 into 24 ( $575 \mathrm{mg}, 84 \%$ ) as a white solid: $[\alpha]_{\mathrm{D}}^{25}=-7.3\left(c 0.6, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 7.64-7.61(\mathrm{~m}, 2 \mathrm{H}), 7.48-7.18(\mathrm{~m}, 29 \mathrm{H}), 7.00-6.98(\mathrm{~m}, 2 \mathrm{H}), 5.60$ $(\mathrm{d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.89-4.79(\mathrm{~m}, 3 \mathrm{H}), 4.41$ $(\mathrm{d}, J=11.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.28-4.22(\mathrm{~m}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H})$, $3.91-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.70-3.63(\mathrm{~m}, 2 \mathrm{H}), 3.58(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.23$ $(\mathrm{dd}, J=4.5,10.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 165.9,150.1$, 148.2, 137.8, 137.7 (2 C), 136.6, 127.8, 127.6, 127.3, 127.2, 127.1 (2 C), 126.7, 126.6, 126.5, 126.4, 125.9, 123.1, 122.4, 114.4, 111.6, 98.7, 85.4, $77.6,76.2,74.5,74.1,73.5,72.1,71.5,70.8,61.8,55.0,51.0$; HRMS (MALDI) calcd for $\mathrm{C}_{55} \mathrm{H}_{52} \mathrm{O}_{9} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$879.3504, found 879.3512.

Methyl 3,5-Dimethoxy-4-(2', $3^{\prime}, 4^{\prime}$-tri-O-benzyl-6'-O-trityl-$\boldsymbol{\beta}$-d-allopyranosyloxy)benzoate (25). Similar procedure as that used for $\mathbf{1 4} \rightarrow \mathbf{1 5}$ was adopted to convert tetraol 23 into $25(29 \mathrm{mg}, 81 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=+38.0\left(c 3.2, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.31-7.10(\mathrm{~m}, 30 \mathrm{H}), 6.91(\mathrm{~d}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 5.73(\mathrm{~d}, J=6.0$ $\mathrm{Hz}, 1 \mathrm{H}), 5.08(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{~d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.72-4.67$ $(\mathrm{m}, 2 \mathrm{H}), 4.29(\mathrm{~d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.13(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 2 \mathrm{H})$, $3.95-3.93(\mathrm{~m}, 1 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 3.73(\mathrm{~s}, 6 \mathrm{H}), 3.62(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, $3.50(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.34(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.10-3.07(\mathrm{~m}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 165.8,152.0,143.0,138.4,138.2,136.9$, $127.8,127.3,127.2,127.0,126.8,126.6,126.5,126.3,126.2,125.7,124.4$, 106.3, 100.7, 85.1, 78.9, 76.2, 74.8, 74.4, 73.3, 72.0, 71.6, 70.4, 61.4, 55.5, 51.2; HRMS (MALDI) calcd for $\mathrm{C}_{56} \mathrm{H}_{54} \mathrm{O}_{10} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 909.3609$, found 909.3623.

3-Methoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}$-tri-O-benzyl-6'-O-trityl- $\beta$-d-allopyranosyloxy)benzoic acid (26). To a solution of the methyl ester 24 ( $583 \mathrm{mg}, 0.68 \mathrm{mmol}$ ) in THF ( 10 mL ) was added dropwise a solution of $\mathrm{KOH}(1.14 \mathrm{~g}, 20.40 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL})$. The mixture was stirred at $70^{\circ} \mathrm{C}$ for 10 h . HOAc was added to adjust the pH value of the reaction mixture to 7.0 . The resulting mixture was extracted with ethyl acetate $(100 \mathrm{~mL} \times 3)$. The combined organic layers were washed with saturated NaCl , dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under vacuum. The residue was purified by column chromatography on silica gel $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}, 50: 1\right)$ to give $26(499 \mathrm{mg}, 87 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=-7.6\left(c 1.2, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.76(\mathrm{dd}$, $J=1.8,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.67(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.49-7.18(\mathrm{~m}, 29 \mathrm{H})$, $7.01-6.98(\mathrm{~m}, 2 \mathrm{H}), 5.63(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~d}, J=12.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.90-4.80(\mathrm{~m}, 3 \mathrm{H}), 4.42(\mathrm{~d}, J=11.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.29-4.22(\mathrm{~m}, 3 \mathrm{H}), 3.95$ $(\mathrm{s}, 3 \mathrm{H}), 3.72-3.65(\mathrm{~m}, 2 \mathrm{H}), 3.61(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.24(\mathrm{dd}, J=4.5$, $10.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 171.2,151.8,149.2,143.9$, 138.7, 137.6, 128.7, 128.5, 128.4, 128.3, 128.2, 128.1, 128.0, 127.9, 127.7, 127.5, 127.4, 127.2, 127.0, 126.8, 124.3, 123.1, 115.3, 112.9, 99.7, 86.3, $78.6,77.2,75.5,75.0,74.4,73.0,72.5,71.8,62.7,55.9$; HRMS (MALDI) calcd for $\mathrm{C}_{54} \mathrm{H}_{50} \mathrm{O}_{9} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$865.3347, found 865.3354.

3,5-Dimethoxy-4-( $2^{\prime}, 3^{\prime}, 4^{\prime}$-tri-O-benzyl- $6^{\prime}$-O-trityl- $\beta$-d-allopyranosyloxy)benzoic acid (27). Similar procedure as that used for $24 \rightarrow 26$ was adopted to convert methyl ester 25 into acid $27(25 \mathrm{mg}$, $92 \%$ ) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=+41.0\left(c 1.8, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR ( 300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43-7.18(\mathrm{~m}, 30 \mathrm{H}), 6.99-6.96(\mathrm{~m}, 2 \mathrm{H}), 5.84(\mathrm{~d}, J=$ $7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{~d}, J=12.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.92(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.80$ (dd, $J=4.5,12.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.37(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.22(\mathrm{~d}, J=11.4 \mathrm{~Hz}$, $2 \mathrm{H}), 4.05-4.01(\mathrm{~m}, 1 \mathrm{H}), 3.82(\mathrm{~s}, 6 \mathrm{H}), 3.71(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.60$ (d, $J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.19(\mathrm{dd}, J=5.1,9.9 \mathrm{~Hz}, 1$ H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 170.6, 152.1, 143.0, 138.9, 138.3, 138.2, 136.8, 127.8, 127.7, 127.5 (2 C), 127.2, 127.0, 126.9, 126.8, 126.6, 126.5, 126.4, 126.2 (2 C), 125.8, 123.5, 106.8, 100.6, 85.1, 78.8, 74.8, 74.4, 73.3, 72.0, 71.6, 70.5, 61.4, 55.5; HRMS (MALDI) calcd for $\mathrm{C}_{55} \mathrm{H}_{52} \mathrm{O}_{10} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$895.3453, found 895.3462.

Dimeric Ester 28. To a solution of alcohol $17(320 \mathrm{mg}, 0.50 \mathrm{mmol})$ and acid $26(843 \mathrm{mg}, 1.00 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15 \mathrm{~mL})$ were added DCC ( $206 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) and DMAP ( $61 \mathrm{mg}, 0.5 \mathrm{mmol}$ ). After being stirred at room temperature for 24 h , the mixture was concentrated under vacuum. The residue was purified by column chromatography on silica gel (petroleum ether/EtOAc, 7:1) to give $28(619 \mathrm{mg}, 76 \%)$ as a white solid: $[\alpha]^{25}{ }_{D}=-14.5\left(c 0.6, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{HNMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.65-6.98$ $(\mathrm{m}, 51 \mathrm{H}), 6.03-5.94(\mathrm{~m}, 1 \mathrm{H}), 5.62(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.36(\mathrm{~d}, J=16.8$ $\mathrm{Hz}, 1 \mathrm{H}), 5.24(\mathrm{~d}, J=10.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.14-5.01(\mathrm{~m}, 3 \mathrm{H}), 4.91-4.82(\mathrm{~m}$, $6 \mathrm{H}), 4.75(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.67(\mathrm{t}, J=11.7 \mathrm{~Hz}, 2 \mathrm{H}), 4.50(\mathrm{dd}, J=3.6$, $12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.42(\mathrm{~d}, J=11.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.30-4.23(\mathrm{~m}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H})$, $3.85(\mathrm{~s}, 3 \mathrm{H}), 3.83-3.70(\mathrm{~m}, 6 \mathrm{H}), 3.63(\mathrm{~d}, J=10.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.23(\mathrm{dd}, J=$ 3.6, $10.2 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 167.7, 167.5, 153.1, 152.2, 151.1, 151.0, 145.7, 140.5, 140.0 (2 C), 139.4, 139.2, 134.1, 130.5, 130.2 (2 C), 130.1 (2 C), 130.0, 129.9, 129.8, 129.7, 129.6, 129.5, 129.4, 129.2, 129.1, 129.0, 128.6, 126.6, 125.4, 125.3, 124.9, 119.8, 117.4, 117.1,
114.6, 114.5, 103.7, 101.5, 88.0, 86.1, 83.3, 80.4, 79.1, 77.6, 77.2, 76.9, 76.8, 76.6, 76.2, 75.2, 74.8, 74.2, 73.6, 67.2, 64.9, 64.4, 57.8, 57.6; HRMS (MALDI) calcd for $\mathrm{C}_{92} \mathrm{H}_{88} \mathrm{O}_{17} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1487.5914, found 1487.5913.

Dimeric Ester 29. Similar procedure as that used for $\mathbf{1 7 + 2 6 \rightarrow 2 8}$ was applied to condense $\mathbf{1 7}$ and 27 into $29(95 \mathrm{mg}, 74 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=+9.3\left(c 1.9, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.49-7.46$ $(\mathrm{m}, 2 \mathrm{H}), 7.33-7.03(\mathrm{~m}, 46 \mathrm{H}), 6.91-6.89(\mathrm{~m}, 2 \mathrm{H}), 5.93-5.86(\mathrm{~m}, 1 \mathrm{H})$, $5.74(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.28(\mathrm{dd}, J=1.2,17.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{dd}, J=3.6$, $10.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.06(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.94(\mathrm{~d}, J=$ $11.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.84-4.55(\mathrm{~m}, 10 \mathrm{H}), 4.43(\mathrm{dd}, J=5.6,12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.28$ $(\mathrm{d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{dd}, J=1.6,13.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.95-3.92(\mathrm{~m}, 1 \mathrm{H})$, $3.75(\mathrm{~s}, 3 \mathrm{H}), 3.70(\mathrm{~s}, 6 \mathrm{H}), 3.75-3.70(\mathrm{~m}, 3 \mathrm{H}), 3.61-3.59(\mathrm{~m}, 1 \mathrm{H}), 3.52$ ( $\mathrm{dd}, J=2.0,11.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.34(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.09(\mathrm{dd}, J=4.8,10.0$ $\mathrm{Hz}, 1 \mathrm{H}){ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 165.0,164.8,152.1,149.6,148.5$, 143.0, 138.7, 138.4, 138.3, 137.3 (2 C), 136.9, 136.6, 131.3, 127.8, 127.6, 127.5, 127.4 (2 C), 127.2, 127.1 (2 C), 127.0 (3 C), 126.8, 126.7, 126.5, 126.4, 126.3, 126.2, 125.8, 124.0 (2 C), 122.2, 117.2, 114.6, 111.9, 106.8, 101.1, 100.7, 85.1, 83.4, 80.6, 79.0, 74.9, 74.8, 74.4, 74.2, 73.9, 73.4, 72.6, 72.0, 71.6, 70.5, 64.6, 62.7, 61.4, 55.8, 54.9; HRMS (MALDI) calcd for $\mathrm{C}_{93} \mathrm{H}_{90} \mathrm{O}_{18} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1517.6019, found 1517.6002.

Compound 30. Similar procedure as that used for $15 \rightarrow 16$ was applied to convert trityl ether 28 into alcohol $30(502 \mathrm{mg}, 81 \%)$ as a white solid: $[\alpha]^{25}=-5.8\left(c \quad 1.3, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.53(\mathrm{dd}, J=1.2,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{dd}, J=1.2,6.4 \mathrm{~Hz}, 2 \mathrm{H})$, 7.43 (dd, $J=1.2,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.16(\mathrm{~m}, 30 \mathrm{H}), 7.07(\mathrm{dd}, J=8.8$, $16.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.97-5.87(\mathrm{~m}, 1 \mathrm{H}), 5.56(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.30(\mathrm{~d}, J=$ $17.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{t}, J=12.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.97-4.92(\mathrm{~m}, 3 \mathrm{H}), 4.87-4.68$ (m, 8H), 4.56-4.46 (m, 3 H), 4.40-4.35 (m, 2 H), 4.14-4.09 (m, 2 H), $3.39(\mathrm{dd}, J=1.5,12.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 6 \mathrm{H}), 3.66-3.60(\mathrm{~m}, 4 \mathrm{H})$, $3.60-3.56(\mathrm{~m}, 1 \mathrm{H}), 3.46-3.35(\mathrm{~m}, 2 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 164.8(2 \mathrm{C}), 150.0,149.5,148.4,148.3,137.8,137.5,137.3$ (2 C), 136.7, 136.6, 131.3, 127.5, 127.4, 127.3, 127.2, 127.1, 127.0, 126.9 (2 C), 126.8 (3 C), 126.6, 126.5, 123.9, 122.9, 122.3, 122.2, 117.1, 115.0, 113.8, 112.0, 111.8, 101.0, 98.4, 83.4, 80.6, 77.3, 76.6, 76.2, 74.8, 74.2, 74.1, 73.8, 73.7, 73.5, 72.5, 72.2, 72.0, 70.6, 64.6, 62.4, 60.9, 55.0, 54.9; HRMS (MALDI) calcd for $\mathrm{C}_{73} \mathrm{H}_{74} \mathrm{O}_{17} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1245.4818, found 1245.4802 .

Compound 31. Similar procedure as that used for $15 \rightarrow 16$ was applied to convert trityl ether 29 into alcohol $31(76 \mathrm{mg}, 90 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-2.4\left(c 0.8, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.58$ $(\mathrm{s}, 1 \mathrm{H}), 7.55(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.26(\mathrm{~m}, 32 \mathrm{H}), 7.10(\mathrm{~d}, J=8.7 \mathrm{~Hz}$, $1 \mathrm{H}), 6.01-5.96(\mathrm{~m}, 1 \mathrm{H}), 5.62(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.39(\mathrm{~d}, J=17.1 \mathrm{~Hz}$, $1 \mathrm{H}), 5.27-5.19(\mathrm{~m}, 2 \mathrm{H}), 5.12-4.62(\mathrm{~m}, 13 \mathrm{H}), 4.57(\mathrm{~d}, J=11.4 \mathrm{~Hz}, 1 \mathrm{H})$, $4.46-4.40(\mathrm{~m}, 2 \mathrm{H}), 4.21(\mathrm{~s}, 1 \mathrm{H}), 3.97-3.93(\mathrm{~m}, 1 \mathrm{H}), 3.86(\mathrm{~s}, 3 \mathrm{H}), 3.80$ $(\mathrm{s}, 6 \mathrm{H}), 3.86-3.80(\mathrm{~m}, 5 \mathrm{H}), 3.67-3.52(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta$ 164.7, 151.9, 149.6, 148.4, 138.4, 138.1, 137.3, 137.2, 136.8, 136.5, 131.3, 127.5, 127.4 (3 C), 127.3, 127.2, 127.1 (2 C), 127.0, 126.8 (3 C), 126.3 (2 C), 124.5, 123.9, 122.1, 117.2, 114.6, 111.9, 106.3, 101.2, 101.0, 83.3, 80.5, 78.7, 76.5, 76.2, 74.8, 74.2 (2 C), 74.0, 73.8, 73.3, 72.6, 72.0, 70.5, 64.5, 62.7, 61.1, 55.4, 54.9; HRMS (MALDI) calcd for $\mathrm{C}_{74} \mathrm{H}_{76} \mathrm{O}_{18} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1275.4924, found 1275.4939.

Alcoholic Acid 32. To the solution of allyl ester 30 ( $104 \mathrm{mg}, 0.09$ $\mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8 \mathrm{~mL})$ and $\mathrm{Et}_{2} \mathrm{NH}(2 \mathrm{~mL})$ was added $\mathrm{Pd}\left(\mathrm{Ph}_{3} \mathrm{P}\right)_{4}$ $(12 \mathrm{mg}, 0.01 \mathrm{mmol})$. The solution was stirred at room temperature for 4 h . Evaporation to remove the solvent afforded a residue, which was purified by column chromatography on silica gel $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right.$, $50: 1)$ to give acid $32(90 \mathrm{mg}, 89 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=+3.1(c 1.2$, $\left.\mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.51(\mathrm{dd}, J=1.6,8.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.45(\mathrm{dd}, J=1.6,8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.35-7.18(\mathrm{~m}, 30 \mathrm{H}), 7.08(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.98(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.59(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~d}, J=10.8$ $\mathrm{Hz}, 1 \mathrm{H}), 4.96-4.69(\mathrm{~m}, 9 \mathrm{H}), 4.59-4.52(\mathrm{~m}, 2 \mathrm{H}), 4.46(\mathrm{~d}, J=11.2 \mathrm{~Hz}$, $1 \mathrm{H}), 4.36(\mathrm{~d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.21-3.11(\mathrm{~m}, 3 \mathrm{H}), 3.89(\mathrm{dd}, J=2.4$, $12.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.76(\mathrm{~s}, 3 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 3.75-3.70(\mathrm{~m}, 4 \mathrm{H})$, $3.52-3.43(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 164.8,150.2$,
149.2, 148.4, 148.3, 137.4, 137.3, 137.2, 136.6, 136.5, 127.6, 127.5 (2 C), 127.4 ( 4 C), 127.2, 127.1, 127.0 (2 C), 126.9, 126.8 (3 C), 126.6, 124.1, 123.0, 122.5, 122.4, 115.3, 113.7, 112.0, 111.9, 100.8, 98.5, 83.5, 80.4, 77.1, 76.2, 74.8, 74.3, 74.1, 73.6, 73.4, 73.4, 72.3, 72.1, 72.0, 70.7, 62.7, 60.9, 55.0, 54.8; HRMS (MALDI) calcd for $\mathrm{C}_{70} \mathrm{H}_{70} \mathrm{O}_{17} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 1205.4505, found 1205.4500.

Alcoholic Acid 33. Similar procedure as that for $30 \rightarrow 32$ was used to convert allyl ester 31 into $33(42 \mathrm{mg}, 87 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=+2.8$ $\left(c 0.9, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.46(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.33-7.16(\mathrm{~m}, 33 \mathrm{H}), 6.94(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.57(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $5.14(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.02(\mathrm{~d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.96-4.75(\mathrm{~m}, 7 \mathrm{H})$, $4.62(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.58(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.48(\mathrm{~d}, J=11.2 \mathrm{~Hz}$, $1 \mathrm{H}), 4.36(\mathrm{~m}, 2 \mathrm{H}), 4.16(\mathrm{~s}, 1 \mathrm{H}), 3.92-3.89(\mathrm{~m}, 1 \mathrm{H}), 3.80-3.74(\mathrm{~m}, 3 \mathrm{H})$, $3.76(\mathrm{~s}, 3 \mathrm{H}), 3.68(\mathrm{~s}, 6 \mathrm{H}), 3.64(\mathrm{dd}, J=4.8,11.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.53-3.50(\mathrm{~m}$, $2 \mathrm{H}), 3.40(\mathrm{dd}, J=9.6,18.0 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 164.6, 151.8, 149.4, 148.4, 138.5, 138.0, 137.7, 137.3, 137.2, 136.6, 136.5, 127.6, 127.5, 127.4 (2 C), 127.2, 127.1, 127.0, 126.9 (2 C), 126.8 (2 C), $126.4,126.3,124.5,122.6,114.8,112.0,106.2,101.4,100.8,83.4,80.4$, $78.5,76.9,76.2,74.8,74.2,74.1,73.8,73.7,73.3,72.5,72.2,72.0,70.5,62.8$, 61.0, 55.4, 54.8; HRMS (MALDI) calcd for $\mathrm{C}_{71} \mathrm{H}_{72} \mathrm{O}_{18} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 1235.4611, found 1235.4609.

Cyclic Dimer 34. To a solution of $32(13 \mathrm{mg}, 0.01 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}$ $(0.08 \mathrm{~mL})$ in dry THF ( 1.00 mL ) was added 2,4,6-trichlorobenzoyl chloride ( $0.05 \mathrm{~mL}, 0.3 \mathrm{mmol}$ ). The solution was stirred at room temperature for 3 h , and then a solution of DMAP ( $86 \mathrm{mg}, 0.72 \mathrm{mmol}$ ) in toluene $(40 \mathrm{~mL})$ was added. The resulting mixture was stirred for another 4 h at the same temperature and was then diluted with ethyl acetate $(300 \mathrm{~mL})$. The resulting mixture was washed with saturated $\mathrm{NaHCO}_{3}$, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under reduced pressure. The residue was subjected to column chromatography on silica gel (petroleum ether/EtOAc, 3:1) to give $34(10 \mathrm{mg}, 78 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=+44.6\left(c 1.1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.68-7.64(\mathrm{~m}, 1 \mathrm{H}), 7.48(\mathrm{dd}, J=2.0,5.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.39-7.17(\mathrm{~m}, 32 \mathrm{H})$, $5.61(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.15(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.06-4.75(\mathrm{~m}, 9 \mathrm{H})$, $4.61-4.43(\mathrm{~m}, 5 \mathrm{H}), 4.34(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $4.09(\mathrm{t}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.01(\mathrm{t}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.88(\mathrm{dd}, J=1.6,9.6$ $\mathrm{Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.83-3.74(\mathrm{~m}, 2 \mathrm{H}), 3.58(\mathrm{dd}, J=2.4$, $8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.44-3.39(\mathrm{~m}, 1 \mathrm{H}), 3.31(\mathrm{dd}, J=2.4,9.6 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 164.8$ (2 C), 149.4, 149.2, 148.4, 148.3, 137.6, 137.4, 137.3, 137.2, 136.5, 136.1, 127.6, 127.5, 127.4, 127.2, 127.1, 127.0, 126.9 (2 C), 126.8 (2 C), 126.7, 126.6, 123.3, 122.7, 122.4, 122.2, 114.2, 114.0, 111.7, 100.1, 97.4, 83.7, 80.2, 77.4, 76.8, 76.2, 74.9, 74.2, 73.9, 73.7, 73.6, 72.3, 72.0, 70.4, 70.0, 64.2, 63.6, 55.0, 54.9; HRMS (MALDI) calcd for $\mathrm{C}_{70} \mathrm{H}_{68} \mathrm{O}_{16} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1187.4400, found 1187.4391.

Cyclic Dimer 35. Similar procedure as that for $32 \rightarrow 34$ was adopted to convert 33 into $35(20 \mathrm{mg}, 88 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-9.5(c 1.2$, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.37-7.14(\mathrm{~m}, 32 \mathrm{H}), 6.82(\mathrm{dd}, J=1.6,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $1 \mathrm{H}), 5.46(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.20(\mathrm{~d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~d}, J=10.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.95(\mathrm{dd}, J=4.4,11.6 \mathrm{~Hz}, 3 \mathrm{H}), 4.86-4.72(\mathrm{~m}, 5 \mathrm{H}), 4.69(\mathrm{~d}, J=11.6$ $\mathrm{Hz}, 1 \mathrm{H}), 4.59(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.51(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.48(\mathrm{dd}, J=$ $3.6,12.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.32(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, 4.13-4.10 (m, 1 H), 3.97-3.87 (m, 2 H), $3.80(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H})$, $3.77-3.72(\mathrm{~m}, 3 \mathrm{H}), 3.64(\mathrm{dd}, J=2.4,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 3.41(\mathrm{t}, J=$ $8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.33(\mathrm{dd}, J=2.4,9.6 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $165.9,165.4,154.1,152.9,150.5,149.1,139.3,139.2,139.1,138.3,138.1$, 137.6, 137.3, 128.6, 128.5 (2 C), 128.4, 128.2, 128.1, 128.0 (2 C), 127.9, 127.8, 127.4, 127.3, 125.7, 124.4, 123.5, 114.5, 112.4, 107.0, 106.8, 103.0, 101.5, 84.5, 81.1, 79.2, 78.6, 77.4, 76.3, 76.0, 75.3, 74.9, 74.8, 74.6, 73.3, 73.1, 71.1, 70.8, 65.3, 63.8, 56.2, 56.1, 55.9; HRMS (MALDI) calcd for $\mathrm{C}_{71} \mathrm{H}_{70} \mathrm{O}_{17} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1217.4505, found 1217.4520.

Clemoarmanoside A (7). To a solution of 34 ( $30 \mathrm{mg}, 0.026$ $\mathrm{mmol})$ in $\mathrm{THF} / \mathrm{MeOH}(5 \mathrm{~mL}, \mathrm{v} / \mathrm{v}=3: 2)$ was added $\mathrm{Pd}(\mathrm{OH})_{2}(3 \mathrm{mg}$, $0.02 \mathrm{mmol})$. The reaction flask was evacuated and refilled with $\mathrm{H}_{2}$. The
mixture was stirred at room temperature for 3 h under $\mathrm{H}_{2}$ atmosphere. Filtration through a pad of Celite and evaporation of the filtrate under vacuum gave a residue, which was purified by column chromatography on silica gel $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}, 6: 1\right)$ to give $7(13 \mathrm{mg}, 81 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=+80.8$ (c0.3, pyridine); ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{SOCD}_{3}\right) \delta$ $7.79(\mathrm{dd}, J=2.0,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{dd}, J=1.5,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.42(\mathrm{~d}, J=2.5$ $\mathrm{Hz}, 1 \mathrm{H}), 7.41(\mathrm{t}, J=3.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{~d}, J=9.5$ $\mathrm{Hz}, 1 \mathrm{H}), 5.52(\mathrm{~s}, 1 \mathrm{H}), 5.45(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.39(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H})$, $5.33(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.25(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.21(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H})$, $5.06(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.42(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.38(\mathrm{~d}, J=11.5 \mathrm{~Hz}$, $1 \mathrm{H}), 4.28-4.24(\mathrm{~m}, 1 \mathrm{H}), 4.13-4.07(\mathrm{~m}, 2 \mathrm{H}), 3.99-3.97(\mathrm{~m}, 2 \mathrm{H}), 3.79$ $(\mathrm{s}, 6 \mathrm{H}), 3.58(\mathrm{~m}, 1 \mathrm{H}), 3.45-3.39(\mathrm{~m}, 3 \mathrm{H}), 3.18-3.16(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{SOCD}_{3}$ ) $\delta 165.2,165.1,150.1,149.8,148.5,148.4$, 123.1, 123.0, 122.5, 122.4, 114.4, 114.3, 112.1, 98.3, 96.8, 76.9, 73.5, 72.8, 71.7, 71.2, 70.6, 69.8, 68.2, 65.4, 65.1, 55.5; HRMS (ESI) calcd for $\mathrm{C}_{28} \mathrm{H}_{32} \mathrm{O}_{16} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$647.1588, found 647.1600.

Clemahexapetoside A (8). Similar procedure as that used for $34 \rightarrow 7$ was applied to convert 35 into Clemahexapetoside A $(8)(20 \mathrm{mg}$, $100 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=+81.1\left(c 0.2\right.$, pyridine); ${ }^{1} \mathrm{H}$ NMR $(500$ $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{SOCD}_{3}\right) \delta 7.50(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.36(\mathrm{~d}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.08(\mathrm{~d}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.99(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{dd}, J=1.5,9.0$ $\mathrm{Hz}, 1 \mathrm{H}), 5.61(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.47(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.43(\mathrm{~d}, J=$ $5.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.29(\mathrm{~d}, J=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.15(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.07(\mathrm{~d}$, $J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.05(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.87(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.51(\mathrm{~d}, J=$ $10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.44-4.39(\mathrm{~m}, 1 \mathrm{H}), 4.36(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.96(\mathrm{~s}$, $3 \mathrm{H}), 3.94-3.83(\mathrm{~m}, 4 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.59(\mathrm{~s}, 3 \mathrm{H}), 3.49-3.46(\mathrm{~m}$, $1 \mathrm{H}), 3.40-3.37(\mathrm{~m}, 3 \mathrm{H}), 3.22-3.19(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 125 MHz , $\left.\mathrm{CD}_{3} \mathrm{SOCD}_{3}\right) \delta 165.2,164.8,153.2,152.0,149.9,148.4,138.1,124.5$, 122.6, 122.3, 114.4, 111.9, 107.0, 106.9, 99.2, 99.0, 77.2, 73.9, 73.0, 71.7, 71.4, 71.0, 70.7, 68.6, 65.4, 64.3, 56.6, 55.9, 55.5; HRMS (ESI) calcd for $\mathrm{C}_{29} \mathrm{H}_{34} \mathrm{O}_{17} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$677.1688, found 677.1707.

Methyl 4-(2', $3^{\prime}, 4^{\prime}$-Tri-O-benzyl-6'-O-trityl- $\beta$-d-glucopyranosyloxy)benzoate (37). Similar procedure as that used for $14 \rightarrow \mathbf{1 5}$ was applied to convert tetraol 36 into 37 ( $438 \mathrm{mg}, 70 \%$ for two steps) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-25.4\left(c 0.9, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.97(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.40-7.09(\mathrm{~m}, 30 \mathrm{H})$, $6.81(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.05(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{~d}, J=10.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.86-4.74(\mathrm{~m}, 3 \mathrm{H}), 4.64(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.29(\mathrm{~d}, J=10.8 \mathrm{~Hz}$, $1 \mathrm{H}), 3.82(\mathrm{~s}, 3 \mathrm{H}), 3.78(\mathrm{t}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.71-3.61(\mathrm{~m}, 2 \mathrm{H})$, $3.53-3.51(\mathrm{~m}, 2 \mathrm{H}), 3.21(\mathrm{dd}, J=5.2,10.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}(100$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 165.7,159.9,142.7,137.3,137.1,136.6,130.6,127.7$, $127.4,127.3,127.2,127.0,126.8$ (2 C), 126.0, 123.4, 115.4, 100.0, 85.6, 83.6, 81.0, 76.7, 75.0, 74.2, 74.1, 74.0, 61.4, 51.0; HRMS (MALDI) calcd for $\mathrm{C}_{54} \mathrm{H}_{50} \mathrm{O}_{8} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$849.3398, found 849.3409.

4 -( $2^{\prime}, 3^{\prime}, 4^{\prime}$-Tri-O-benzyl-6'-O-trityl- $\beta$-d-glucopyranosyloxy)benzoic acid (38). Similar procedure as that used for $\mathbf{2 4} \boldsymbol{\rightarrow} \mathbf{2 6}$ was applied to convert methyl ester 37 into acid 38 ( $348 \mathrm{mg}, 91 \%$ ) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-25.6\left(c 1.8, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $8.05(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.39-7.10(\mathrm{~m}, 30 \mathrm{H}), 6.82-6.80(\mathrm{~m}, 2 \mathrm{H})$, $5.08(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.00(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.87-4.75(\mathrm{~m}, 3 \mathrm{H})$, $4.65(\mathrm{~d}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.30(\mathrm{~d}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{t}, J=8.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.71-3.64(\mathrm{~m}, 2 \mathrm{H}), 3.55-3.53(\mathrm{~m}, 2 \mathrm{H}), 3.21-3.20(\mathrm{~m}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.6,160.5,142.7,137.3,137.1,136.5$, 131.3, 127.7, 127.4, 127.3, 127.2, 127.0, 126.9, 126.8, 126.0, 122.6, 115.4, 99.9, 85.6, 83.5, 81.0, 76.7, 75.0, 74.2, 74.1, 74.0, 61.3; HRMS (MALDI) calcd for $\mathrm{C}_{53} \mathrm{H}_{48} \mathrm{O}_{8} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$835.3241, found 835.3251.

4-( $2^{\prime}, 3^{\prime}, 4^{\prime}$-Tri-O-benzyl- $\boldsymbol{\beta}$-d-glucopyranosyloxy)benzoic acid (39). Similar procedure as that used for $15 \rightarrow 16$ was applied to convert trityl ether 38 into $39(354 \mathrm{mg}, 80 \%)$ as a white solid: $[\alpha]_{\mathrm{D}}^{25}=-20.5\left(c 0.7, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.01(\mathrm{~d}, J$ $=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.29-7.18(\mathrm{~m}, 15 \mathrm{H}), 6.98(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.10(\mathrm{~d}, J=$ $7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{dd}, J=4.8,10.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.83-4.74(\mathrm{~m}$, $3 \mathrm{H}), 4.62(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.85-3.82(\mathrm{~m}, 1 \mathrm{H}), 3.72-3.57(\mathrm{~m}$, $4 \mathrm{H}), 3.49-3.48(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.9,161.2$,
138.3, 137.9, 137.8, 132.4, 128.6, 128.4, 128.2, 128.1, 128.0, 127.9 (2 C), 127.8, 123.6, 115.9, 100.5, 84.3, 81.8, 75.8, 75.6, 75.2 (2 C), 61.8; HRMS (MALDI) calcd for $\mathrm{C}_{34} \mathrm{H}_{34} \mathrm{O}_{8} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 593.2146$, found 593.2150 .

Allyl 4-(2', 3', $4^{\prime}$-Tri-O-benzyl- $\beta$-d-glucopyranosyloxy)benzoate (40). Similar procedure as that used for $16 \rightarrow \mathbf{1 7}$ was applied to convert acid 39 into allyl ester $40(308 \mathrm{mg}, 81 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-11.8$ $\left(c 0.4, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.05(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.36-7.30(\mathrm{~m}, 15 \mathrm{H}), 7.03(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.07-6.00(\mathrm{~m}, 1 \mathrm{H})$, $5.43-5.38(\mathrm{~m}, 1 \mathrm{H}), 5.30-5.27(\mathrm{~m}, 1 \mathrm{H}), 5.16(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, $4.99-4.80(\mathrm{~m}, 7 \mathrm{H}), 4.69(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{dd}, J=2.4,12.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.80-3.64(\mathrm{~m}, 4 \mathrm{H}), 3.56-3.53(\mathrm{~m}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 164.7,159.6,137.4,136.9,136.8,131.4,130.8,127.6,127.5,127.2$, 127.1 (2 C), 127.0, 126.9, 126.8, 123.6, 117.2, 114.9, 99.6, 83.4, 80.9, 74.8, 74.6, 74.3, 74.2, 64.5, 60.9; HRMS (MALDI) calcd for $\mathrm{C}_{37} \mathrm{H}_{38} \mathrm{O}_{8} \mathrm{Na}$ [M + $\mathrm{Na}]^{+} 633.2459$, found 633.2459 .

Dimeric Ester 41. To a solution of alcohol $40(400 \mathrm{mg}, 0.65 \mathrm{mmol})$ and acid $38(800 \mathrm{mg}, 0.98 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ were added $\operatorname{EDCI}(200 \mathrm{mg}, 1.30 \mathrm{mmol})$ and DMAP ( $79 \mathrm{mg}, 0.65 \mathrm{mmol}$ ). After being stirred at room temperature for 24 h , the mixture was concentrated under vacuum. The residue was purified by column chromatography on silica gel (petroleum ether/EtOAc, 8:1) to provide $41(800 \mathrm{mg}, 87 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-27.2\left(c 0.5, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{HNMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.94(\mathrm{~d}, J=$ $8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.90(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.37-7.10(\mathrm{~m}, 45 \mathrm{H}), 6.97(\mathrm{~d}, J=8.8$ $\mathrm{Hz}, 2 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.92-5.83(\mathrm{~m}, 1 \mathrm{H}), 5.27(\mathrm{dd}, J=1.2$, $16.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.15-4.74(\mathrm{~m}, 12 \mathrm{H}), 4.65-4.55(\mathrm{~m}, 5 \mathrm{H}), 4.33-4.27(\mathrm{~m}$, $2 \mathrm{H}), 3.80-3.51(\mathrm{~m}, 9 \mathrm{H}), 3.20(\mathrm{dd}, J=4.4,10.0 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 164.6, 160.2, 159.6, 142.7, 137.4, 137.2, 136.9, 136.7, 136.4, 131.3, 130.8, 130.6, 127.7, 127.5 (2 C), 127.4, 127.2 (2 C), 127.1, 127.0 (3 C), 126.9 (2 C), 126.8 (2 C), 126.7, 126.0, 123.6, 122.9, 117.0, 115.4, 115.1, 99.9, 99.8, 85.5, 83.6, 81.1, 80.8, 76.8, 75.0, 74.9, 74.2, 74.1 (2 C), 73.9, 72.6, 64.3, 62.1, 61.3; HRMS (MALDI) calcd for $\mathrm{C}_{90} \mathrm{H}_{84} \mathrm{O}_{15} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 1427.5702, found 1427.5747.

Trimeric Ester 42. Similar procedure as that used for $\mathbf{4 0}+\mathbf{3 8} \boldsymbol{\rightarrow 4 1}$ was applied to condense acid 38 and the alcohol resulting from trityl ether 41 into $42(275 \mathrm{mg}, 76 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-40.5(c 0.5$, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.98-7.86(\mathrm{~m}, 6 \mathrm{H}), 7.42-6.85$ $(\mathrm{m}, 66 \mathrm{H}), 6.02-5.87(\mathrm{~m}, 1 \mathrm{H}), 5.35-4.63(\mathrm{~m}, 26 \mathrm{H}), 4.38-4.31(\mathrm{~m}, 3 \mathrm{H})$, $4.20(\mathrm{dd}, J=8.8,15.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.04-3.54(\mathrm{~m}, 13 \mathrm{H}), 3.29-3.24(\mathrm{~m}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 164.6$ (2 C), 164.3, 160.1, 159.9, 159.4, 142.7, 142.6, 137.4, 137.2 (2 C), 137.0, 136.9, 136.6, 136.5, 131.3, 130.7, 130.6 (2 C), 130.5, 127.7, 127.5, 127.4 (2 C), 127.2 ( 4 C), 127.1, 127.0 (2 C), 126.9, 126.8 (2 C), 126.7, 126.5, 126.0, 123.4, 123.0, 122.7, 116.9, 115.4, 115.1, 115.0, 114.8, 99.8. 99.6, 99.5, 85.6, 83.7, 83.6, 83.5, 81.1, 80.8, 77.0, 76.9, 76.5. 75.1, 75.0, 74.8. 74.2 (2 C), 74.1, 73.7, 72.7, 72.4, 64.3, 62.2, 62.1, 61.3; HRMS (MALDI) calcd for $\mathrm{C}_{124} \mathrm{H}_{116} \mathrm{O}_{22} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 1979.7850, found 1979.7825.

Trimer Derivative 43. Similar procedure as that used for $15 \rightarrow \mathbf{1 6}$ was applied to convert trityl ether 42 into alcohol 43 ( $33 \mathrm{mg}, 85 \%$ ) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-29.8\left(c \quad 0.4, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H} \operatorname{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.87(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.78(\mathrm{dd}, J=5.6,8.4 \mathrm{~Hz}, 4 \mathrm{H})$, $7.29-7.13(\mathrm{~m}, 45 \mathrm{H}), 7.03(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.89(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $6.80(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.92-5.80(\mathrm{~m}, 1 \mathrm{H}), 5.26(\mathrm{dd}, J=1.2,17.2 \mathrm{~Hz}$, $1 \mathrm{H}), 5.17-5.13(\mathrm{~m}, 2 \mathrm{H}), 5.05(\mathrm{t}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.98-4.56(\mathrm{~m}$, $21 \mathrm{H}), 4.40(\mathrm{~d}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.23-4.13(\mathrm{~m}, 2 \mathrm{H}), 3.96-3.92(\mathrm{~m}, 1 \mathrm{H})$, $3.83-3.50(\mathrm{~m}, 13 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 164.6,164.5$, 164.4, 160.0, 159.7, 159.4, 137.4, 137.2 (2 C), 137.0 (2 C), 136.9, 136.6, 136.4, 131.3, 130.6 (2 C), 130.5 (3 C), 127.5 (2 C), 127.4 (2 C), 127.2 (2 C), 127.1 (2 C), 127.0 (3 C), 126.9, 126.8 (2 C), 126.7 (2 C), 123.5, 123.0, 122.8, 116.9, 115.2, 114.8, 114.6, 99.6 (2 C), 99.5, 83.6, 81.0, 80.9, 80.8, 77.0, 76.9, 76.5, 74.9 (2 C), 74.6, 74.2, 74.1, 72.6, 72.2, 64.3, 62.6, 62.5, 60.7; HRMS (MALDI) calcd for $\mathrm{C}_{105} \mathrm{H}_{102} \mathrm{O}_{22} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 1737.6755, found 1737.6759.

Cyclic Trimer 44. Similar procedure as that used for $30 \rightarrow 32 \rightarrow 34$ was applied to convert 43 into 44 ( $12 \mathrm{mg}, 48 \%$ for two steps) as a white
solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-24.4\left(c 0.1, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.85(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 6 \mathrm{H}), 7.27-7.18(\mathrm{~m}, 45 \mathrm{H}), 6.97(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 6 \mathrm{H})$, $5.10(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 4.91-4.87(\mathrm{~m}, 9 \mathrm{H}), 4.80-4.72(\mathrm{~m}, 6 \mathrm{H})$, $4.66-4.57(\mathrm{~m}, 6 \mathrm{H}), 4.29(\mathrm{dd}, J=5.2,11.6 \mathrm{~Hz}, 3 \mathrm{H}), 3.76-3.65(\mathrm{~m}$, $9 \mathrm{H}), 3.55-3.51(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 164.7$, 159.4, 137.2, 136.9, 136.6, 130.6, 127.5 (2 C), 127.4, 127.2, 127.1, 126.9 (2 C), 126.8, 123.3, 115.8, 99.4, 83.4, 81.2, 76.4, 76.2, 74.8, 74.2, 74.0, 73.1, 62.2; HRMS (MALDI) calcd for $\mathrm{C}_{102} \mathrm{H}_{96} \mathrm{O}_{21} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 1679.6336, found 1679.6372.

Cyclic Trimer 9. Similar procedure as that used for $34 \rightarrow 7$ was applied to convert 44 into $9(12 \mathrm{mg}, 100 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-33.0(c 2.3, \mathrm{MeOH}) ;{ }^{1} \mathrm{H}$ NMR ( $\left.400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 7.83$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 6 \mathrm{H}), 7.03(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 6 \mathrm{H}), 4.99(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$, $4.75-4.70(\mathrm{~m}, 3 \mathrm{H}), 4.36(\mathrm{dd}, J=5.6,12.0 \mathrm{~Hz}, 3 \mathrm{H}), 3.64-3.60(\mathrm{~m}$, $3 \mathrm{H}), 3.41-3.29(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 167.4,162.4$, 132.4, 125.3, 117.9, 101.2, 77.6, 76.8, 75.1, 71.6, 64.6; HRMS (MALDI) calcd for $\mathrm{C}_{39} \mathrm{H}_{42} \mathrm{O}_{21} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 869.2111$, found 869.2145 .

Tetramer 45. Similar procedure as that used for $\mathbf{4 0}+38 \rightarrow 41$ was applied to condense acid 38 and alcohol 43 into $45(110 \mathrm{mg}, 78 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-71.6\left(c 0.7, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.73-6.72(\mathrm{~m}, 91 \mathrm{H}), 5.85-5.75(\mathrm{~m}, 1 \mathrm{H}), 5.23-4.43(\mathrm{~m}, 32$ H), 4.20-3.18 (m, 24 H); $\left.{ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{( } 100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 164.7, 164.3, 164.1 (2 C), 159.8, 159.6, 159.1, 142.6, 137.5 (2 C), 137.4 (2 C), 137.3, 137.2, 137.0, 136.9, 136.7, 136.5, 136.4 (2 C), 131.4, 130.4 (2 C), 130.3 (2 C), 127.6, 127.5, 127.4 (3 C), 127.3 (2 C), 127.2 (3 C), 127.0, 126.9 (2 C), 126.8 (2 C), 126.7,126.6, 125.9, 123.3, 123.0, 122.6, 122.4, 116.9, 114.8 (2 C), 114.5, 114.4, 99.5, 99.4, 99.1, 85.6, 83.9 (2 C), 83.7, 83.6, 81.1, 81.0, 77.5, 77.4, 77.3 (2 C), 75.2 (2 C), 74.8, 74.7, 74.6, 74.2, 74.1, 74.0, 73.8, 73.5, 73.0, 72.5, 72.2, 64.3, 62.6, 62.3, 62.2, 61.3; HRMS (MALDI) calcd for $\mathrm{C}_{158} \mathrm{H}_{148} \mathrm{O}_{29} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 2531.9998$, found 2532.0021.

Tetrameric Derivative 46. Similar procedure as that used for $15 \rightarrow 16$ was applied to convert trityl ether 45 into alcohol 46 ( 438 mg , $77 \%$ ) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-42.9\left(c 1.1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.69-7.59(\mathrm{~m}, 8 \mathrm{H}), 7.29-7.06(\mathrm{~m}, 60 \mathrm{H}), 6.91(\mathrm{~d}, J=$ $8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.83(\mathrm{~d}, \mathrm{~J}=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.74(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.57(\mathrm{~d}$, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.86-5.76(\mathrm{~m}, 1 \mathrm{H}), 5.24-4.44(\mathrm{~m}, 33 \mathrm{H}), 4.25-3.43$ $(\mathrm{m}, 23 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 164.5$ (2 C), 164.3, 164.1, 159.8, 159.6, 159.4, 159.2, 137.4 (2 C), 137.2, 137.1, 137.0, 136.9 (2 C), 136.7, 136.5, 136.4, 131.3, 130.4, 130.3, 130.2, 127.6, 127.5 (2 C), 127.4 (3 C), 127.3, 127.2 (2 C), 127.1, 127.0 (3 C), 126.9, 126.8, 126.7 (3 C), 126.6, 123.4, 123.1, 122.7, 122.4, 116.9, 114.7, 114.5 (2 C), 114.3, 99.7, 99.4, 99.1, 83.9, 83.8, 83.7, 81.2 (2 C), 81.0, 77.7, 77.5, 76.7, 75.2, 75.0, 74.8 (2 C), 74.6, 74.3, 74.2, 74.1, 73.9, 72.9, 72.2, 71.9, 64.3, 62.9 (2 C), 62.4, 60.6; HRMS (MALDI) calcd for $\mathrm{C}_{139} \mathrm{H}_{134} \mathrm{O}_{29} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 2289.8903, found 2289.8890.

Cyclic Tetramer 47. Similar procedure as that used for $30 \rightarrow$ $32 \rightarrow 34$ was applied to convert 46 into 47 ( $24 \mathrm{mg}, 48 \%$ ) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-24.8\left(c 0.5, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.84(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 8 \mathrm{H}), 7.29-7.16(\mathrm{~m}, 60 \mathrm{H}), 6.96(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 8 \mathrm{H})$, $5.15(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 4.97-4.90(\mathrm{~m}, 8 \mathrm{H}), 4.80-4.76(\mathrm{~m}, 16 \mathrm{H})$, $4.56(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 4 \mathrm{H}), 3.95-3.74(\mathrm{~m}, 16 \mathrm{H}), 3.56-3.51(\mathrm{~m}, 4 \mathrm{H})$; ${ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 164.5,159.8,137.3,137.0,136.5,137.5$, 127.5 (2 C), 127.4, 127.3, 127.2, 127.1, 126.9, 126.8, 122.8, 114.8, 99.3, 83.7, 81.0, 76.6, 74.9, 74.2, 74.0, 72.2, 63.0; HRMS (MALDI) calcd for $\mathrm{C}_{136} \mathrm{H}_{128} \mathrm{O}_{28} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$2231.8484, found 2231.8488 .

Cyclic Tetramer 10. Similar procedure as that used for $34 \rightarrow 7$ was applied to convert $\mathbf{4 7}$ into $\mathbf{1 0}(8 \mathrm{mg}, 100 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-24.1$ (c 0.4, DMF); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{SOCD}_{3}$ ) $\delta 7.78(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 8$ H), 7.06 (d, $J=8.4 \mathrm{~Hz}, 8 \mathrm{H}), 5.21(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 4 \mathrm{H}), 5.09(\mathrm{~d}, J=10.8 \mathrm{~Hz}$, $4 \mathrm{H}), 4.01(\mathrm{t}, \mathrm{J}=10.0 \mathrm{~Hz}, 4 \mathrm{H}), 3.86(\mathrm{t}, J=9.6 \mathrm{~Hz}, 4 \mathrm{H}), 3.45-3.28(\mathrm{~m}$, $12 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{SOCD}_{3}\right) \delta$ 165.1, 160.6, 130.8, 122.6, 115.6, 98.4, 76.4, 73.9, 73.1, 70.7, 64.2; HRMS (ESI) calcd for $\mathrm{C}_{52} \mathrm{H}_{56} \mathrm{O}_{28} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+}$1151.2850, found 1151.2903.

Pentamer 48. Similar procedure as that used for $\mathbf{4 0}+38 \rightarrow 41$ was applied to condense acid 38 with alcohol 46 into ester 48 ( $58 \mathrm{mg}, 76 \%$ ) as a white solid: $[\alpha]_{\mathrm{D}}{ }^{25}=-33.4\left(c 0.7, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.61-7.45(\mathrm{~m}, 6 \mathrm{H}), 7.36-6.82(\mathrm{~m}, 94 \mathrm{H}), 6.72(\mathrm{~d}, J=6.8$ $\mathrm{Hz}, 2 \mathrm{H}), 6.62-6.57(\mathrm{~m}, 6 \mathrm{H}), 6.51(\mathrm{~J}=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.82-5.71(\mathrm{~m}$, $1 \mathrm{H}), 5.27(\mathrm{~d}, J=7.2,1 \mathrm{H}), 5.15-3.21(\mathrm{~m}, 68 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 164.6,164.2,164.1,164.0$ (2 C), 159.8 (2 C), 159.7, 159.5, 158.9, 142.6, 137.7, 137.6, 137.5, 137.4 (2 C), 137.3 (2 C), 137.1, 136.9, $136.8,136.6,136.5,131.4,130.3,130.2$ (2 C), 130.1, 129.9, 128.0, 127.9, $127.6,127.5,127.4(2 \mathrm{C}), 127.3$ (3 C), 127.2 (2 C), 127.1 (3C), 126.9 (3C), 126.8, 126.7, 126.6 (2 C), 126.5, 126.4 (2 C), 126.1, 126.0, 125.8, 123.1, 123.0, 122.8, 122.5, 122.4, 122.2, 116.7, 114.7, 114.6, 114.5, 114.3, 99.8. 99.7, 99.5, 99.1, 85.8, 84.2 (2 C), 84.1, 84.0, 83.9, 83.8, 81.7, 81.6, 81.3, 78.7, 78.1, $77.9,77.4,75.4$ (2 C), 75.2, 75.1, 75.0 (2 C), 74.9, 74.8, 74.7.74.6, 74.5 (2 C), 74.3, 74.2 (2 C), 74.1, 74.0 (2 C), 73.8, 72.7, 72.3. 72.2, 72.0, 64.2, 63.0, 62.8, 62.6, 62.4, 61.3; HRMS (MALDI) calcd for $\mathrm{C}_{192} \mathrm{H}_{180} \mathrm{O}_{36} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 3084.2147, found 3084.2136 .

Pentameric Derivative 49. Similar procedure as that used for $15 \rightarrow 16$ was applied to convert trityl ether 48 into alcohol 49 ( 51 mg , $95 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-16.9\left(c 0.3, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H} \operatorname{NMR}(400$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.58(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{t}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H})$, $7.35-6.85(\mathrm{~m}, 81 \mathrm{H}), 6.63-6.60(\mathrm{~m}, 4 \mathrm{H}), 6.47(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $6.39(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.80-5.70(\mathrm{~m}, 1 \mathrm{H}), 5.29(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, $5.15-3.32(\mathrm{~m}, 68 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 164.6,164.4,164.1$ (2 C), 163.9, 159.8, 159.7, 159.4, 159.3, 158.9, 137.7, 137.5 (2 C), 137.3, 137.2, 137.1, 136.9, 136.8, 136.5 (2 C), 136.4, 131.4, 130.2, 130.1, 130.0, 129.8, 127.5, 127.4 (2 C), 127.3 (2 C), 127.2, 127.1 (2 C), 127.0, 126.9 (2 C), 126.8 ( 3 C), 126.7 ( 2 C), $126.6,126.5,126.1,123.1,123.0,122.9$, 122.6, 122.2, 116.8, 114.7, 114.6, 114.4, 114.2, 114.1, 99.7, 99.4, 99.1. 84.1, 84.0, 83.8, 81.8, 81.6, 81.3, 81.2, 81.1, 78.8, 78.3, 78.1, 77.8, 77.1, 75.2, 75.1, 75.0, 74.9, 74.8, 74.6, 74.4 (2 C), 74.3 (2 C), 74.2 (2 C), 74.1, 74.0, 72.6, 72.2, 71.8 (2 C), 64.1, 63.2, 63.1, 62.9, 62.5, 60.5; HRMS (MALDI) calcd for $\mathrm{C}_{173} \mathrm{H}_{166} \mathrm{O}_{36} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$2842.1051, found 2842.1054.

Cyclic Pentamer 50. Similar procedure as that used for $30 \rightarrow 32 \rightarrow$ 34 was applied to convert 49 into $\mathbf{5 0}(15 \mathrm{mg}, 40 \%$ for two steps) as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=-16.9\left(c \quad 0.3, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.73(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 10 \mathrm{H}), 7.27-7.13(\mathrm{~m}, 80 \mathrm{H}), 6.84(\mathrm{~d}, J=8.8$ $\mathrm{Hz}, 10 \mathrm{H}), 5.06(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 5 \mathrm{H}), 4.92(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 10 \mathrm{H}), 4.80-4.71$ $(\mathrm{m}, 15 \mathrm{H}), 4.58(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 5 \mathrm{H}), 4.51(\mathrm{~d}, J=11.2 \mathrm{~Hz}, 5 \mathrm{H}), 4.17(\mathrm{dd}$, $J=6.8,12.0 \mathrm{~Hz}, 5 \mathrm{H}), 3.78-3.67(\mathrm{~m}, 15 \mathrm{H}), 3.51-3.46(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 164.4,159.6,137.3,137.0,136.7,130.6$, 127.6, 127.5 (2 C), 127.4, 127.3, 127.2, 127.0 (2 C), 126.9, 126.7, 123.0, 115.1, 99.4, 83.6, 80.8, 76.7, 74.9, 74.1, 72.7, 62.3; HRMS (MALDI) calcd for $\mathrm{C}_{170} \mathrm{H}_{160} \mathrm{O}_{35} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$2784.0632, found 2784.0680.

Cyclic Pentamer 11. Similar procedure as that used for $34 \rightarrow 7$ was applied to convert 50 into $\mathbf{1 1}(11 \mathrm{mg}, 100 \%)$ as a white solid: $[\alpha]^{25}{ }_{\mathrm{D}}=13.9$ ( c0.3, MeOH); ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 7.56(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 10 \mathrm{H})$, $6.80(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 10 \mathrm{H}), 5.01(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 5 \mathrm{H}), 4.71-4.68(\mathrm{~d}, J=12.0$ $\mathrm{Hz}, 5 \mathrm{H}), 4.17(\mathrm{dd}, J=7.6,11.6 \mathrm{~Hz}, 5 \mathrm{H}), 3.75(\mathrm{t}, J=8.0 \mathrm{~Hz}, 5 \mathrm{H}), 3.48-3.47$ $(\mathrm{m}, 10 \mathrm{H}), 3.40-3.37(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 167.3$, 162.3, 132.4, 124.8, 117.1, 100.6, 77.9, 76.1, 74.9, 72.0, 65.1 (2 C); HRMS (ESI) calcd for $\mathrm{C}_{65} \mathrm{H}_{70} \mathrm{O}_{35} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$1433.3590, found 1433.3622.

## ■ ASSOCIATED CONTENT

(S) Supporting Information. Additional experimental details, the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra for all new compounds, and the crystallographic data for compound 8 . This material is available free of charge via the Internet at http://pubs.acs.org.

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